Percentage of Revision of B.Sc., Chemistry Syllabus under Regulation 2020

	DEPARTMENT OF CHEMISTRY								
S. No	Course Code	Course Title	Change Details	% of Cha	Content Deleted	Content Updated	Reason for Change (Subject Expert / Course Teacher / Alumini Opinion)		
1	20UCY101	Core Paper – I : Inorganic and Organic Chemistry	Unit 1 - 5	5%	Dehalogenatio n, HIO4	Nomenclature of organic compounds - IUPAC naming of simple and substituted aliphatic, aromatic and alicyclic compounds - priorities of functional groups in polyfunctional compounds.	Subject Expert		
2	20UCY202	Core Paper – II: Organic and Physical Chemistry	Unit 1 - 5	5%	UNIT -II NH2OH, H2NCONHNH2, CrO3, PCC, PDC. UNIT -III Wave	UNIT -III Atomic Structure Rutherford atomic model – Bohr the hydrogen atom – Sommerfeld theo Particle and wave	Subject ory of Expert		

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					mechanics:	character of elec	trons
					Characteristics of wave motion.		1
3	20UCY203	Core Practical – I : Inorganic Qualitative Analysis	Unit 1 - 5	0%	-	-	-
4.	20UCY304	Core - III : Inorganic and Physical Chemistry	Unit 1 - 5	10%	Unit II: Inorganic Polymers: Silicones Preparation, properties and applications. Glass: Manufacture, types and coloured glass. Cement: Types of cement, composition, manufacture and setting of cement. Unit 1: Fertilizers: Important nutrients and their functions. Primary and secondary nutrients. Manufacture of	Unit 1: Extraction and u Platinised asbes Cement: Types o Unit 2: Fertilizers: Impo urea, ammonium nitrate. 5.	tos, colloidal por cement, composition subject

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Ja -							
					urea, ammonium sulphate, super phosphate of lime, triple super phosphate and potassium nitrate.		
5	20UCY3N1	Non Major Elective - I : Food Science and Technology	Unit 1 - 5	0%	- -	-	-
6	20UCY3N2	Non Major Elective - I : Chemistry of Consumer Products	Unit 1 - 5	0%	-	-	-
7	20UCY405	Core - IV : Inorganic, Organic and Physical Chemistry	Unit 1 - 5	10%	Unit I: Extraction and uses of Ti,V,Mo,W and Co. Platinum Metals: Metallurgy of platinum. Platinum black, Platinised asbestos, colloidal platinum - preparation and uses.Group	Unit V: Nuclear Chemistry: Radioactivity- types of radioactive rays -nuclear stability-n/p ratio -magic numbers- nuclear binding energy- mass defect-/nuclear	Course

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Discussions: (i)	shell model -
Cr, Mo and W	groupsdisplace
(iii) Fe,Co and	mentlaw -
Ni . Alloy	decay constant
steels. Heat	- half- life
treatment of	period -
steel. Iron and	radioactive
steel industry in	equilibrium-
India.	transmutation-
Preparation and	artificial
uses of the	transmutation-
following	applications of
compounds.	artificial
TiO2, TiCl4,	transmutation-
CrO2Cl2,	radioactive
ZrOCl2,	series. Nuclear
V2O5,FeSO4,	reactions types:
(NH)4 MoO4,	fission and
PtCl4.	fusion
	reactions-
	principle and
	working of
	nuclear
	reactors.
	Isotopes:
	Separation of
	isotopes-
	identification of
	.isotopes-
	isotopes of
	hydrogen-
	isotope effect-
	application of

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						isotopes in chemistry, agriculture and medicine.	
8	20UCY406	Core Practical II: Volumetric and Organic Qualitative Analysis	Unit 1 - 5	0%			-
9	20UCY4N3	Non Major Elective - II : Water and Water Treatment Processes	Unit 1 - 5	0%			
10	20UCY4N4	Non Major Elective - II : Diagnostic Chemistry	Unit 1 - 5	0%	-	V. 10.	
11	20UPS3A3/ 20UBY3A4 /20UZY3A4	Allied Chemistry Paper – I : Inorganic, Organic and Physical Chemistry	Unit 1 - 5	10%	hybridization, geometry of org molecules – CH C2H4, C2H2, C6 Inductive, Electrometric, Mesomeric,	4, compounds - H6.Nomenclature o heterocyclic compounds- syn and reactions of ve pyrrole- thiophe ts. pyridine.	f Subject Expert thesis furan- ne and

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					law. Determina	conductance tionmeasurements ationdetermination ctandegree of diss of weak electr Ionic product and conductor titrations.	of ociation olyte. of water
12	20UPS4A4/ 20UBY4A5 /20UZY4A5	Allied Chemistry Paper – II : Inorganic, Organic and Physical Chemistry	Unit 1 - 5	5%	s and uses. UNIT-V Emulsions and ge	UNIT-I Cement: Manua – Wet Process ertieprocess- types setting of ceme UNIT-V els:Phase rule- Deferties terms involve phase diagram of	and Dry ent inition
13	20UBY4A6 /20UZY4A6	Allied Chemistry Practical	Unit 1 - 5	0%	-	-	- 1
14	20UCY507	Core Paper – V : Nuclear and Co - ordination Chemistry	Unit 1 - 5	25%	Nuclear and Co-ordination Chemistry Unit-I Radio activity. Mass defect, packing fraction. Nuclear binding energy, n/p ratio and	Coordination and Bioinorganic Chemistry Unit-I Double salts- complex compounds- complex ion and coordination	Course Teacher

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_	stability of the	number-
	nucleus.	Ligands and
	Differences	their
	between	classification-
	nuclear and	chelates and
	chemical	their uses-
	reactions. Half-	coordination
	life period.	number and
	Applications of	stereochemistry
	artificial	of complexes-
	radioactivity:	IUPAC
	Age of the earth	Nomenclature
	and C14 dating,	of coordination
	medical field	compounds.
	applications	Isomerism:
	Isotopes:	ionization,
	Detection of	hydrate, ligand,
	isotope by	linkage,
	Aston Mass	coordination,
	Spectrograph.	position,
	Separation of	geometrical and
	isotopes	optical
	by	isomerisms.
	Electromagneti	Unit-II
	c,	Theories of
	Thermal	coordination
	diffusion and	compounds:
	Electrolytic	Werner's
	techniques.	theory-
	Nuclear fission	Sidwick's
	and fusion.	electronic
	Types of	interpretation-
	nuclear wastes	EAN concept-

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Service The Control of the	aı	nd different	valence bond	
Mark Breedy	ur ur	nethods of	theory outer	
100	d	isposal of	and inner	
	H	ligh and low	orbital	
enge to yell	ra	adioactive	complexes-	
April April 4	, v	vastes.	Limitations of	
and the last a last of the		nit-II	VBT- crystal	
4	c i	Co-ordination	field theory-	
Company Transmit	c	hemistry:	Crystal field	
	e.	Double salt and	splitting in	
	C	oordination	octahedral,	
The transfer of the second	C	compounds.	tetragonal,	
Will Complete the State of the	I	Definitions of	square planar	
Relation by a 4-1-	- t	he terms:	and tetrahedral	
· · · · · · · · · · · · · · · · · · ·	2	Ligands and	complexes-	
	331	Co-ordination	High spin and	
figure 11 and 12 hours at		number.	Low spin	
		Classification	complexes.	
The second second	Dec la la	of ligands.	Unit-III	
		Nomenclature	Crystal field	
整計 下面 有 5页。	and i	of Co-	stablisation	
· · · · · · · · · · · · · · · · · · ·	-1	ordination	energy-	
Control of the Contro	ing to	compounds.	calculation and	
Marie Marie Control		Theories of	uses-	
AL SECTION	× X	Coordination	Limitations of	
ACCEPTANCE OF THE PARTY OF THE	711	compounds:	crystal field	
Agrange had	46,	Werner's	theory.	1000
		Theory:	Applications of	
the second second	- 1	Designation of	coordination	
A STATE OF S	(14)	Cobalt (III)-	compounds:	
	(No.	ammine	Applications of	
		complexes.	copper and	
	Pro-	Sidgwick's	silver	
mana and an annual and an				

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	Theory:	complexes in
	Electronic	qualitative
	interpretation of	analysis.
	Co-ordination	Applications of
	bond - Effective	Ca-EDTA and
	atomic number	Ni-DMG
	rule. VB	complexes in
	theory:	quantitative
	Postulates and	analysis.
	its applications	Unit-IV
,	in the	Properties of
	determination	complexes:
	of geometry	Color of
	and magnetic	transition metal
	property of the	complexes-
	complexes.	visible
	Unit-III	spectrum of
	Spectrochemica	aqueous Ti (III)
	<u>F</u>	ion. Stability of
	series. Chelates:	complexes-
	Definition,	overall and
	classification,	stepwise
	stability factors	formation
	,sequestration	constants.
	and	Factors
	sequestering	affecting
	agents.	stability-
	Detection and	Determination
	structure	of stability
	determination	constant by
	of Complexes:	Job's and
	Solubility	Bjeruum's
	method, change	method.]
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7	in colour,pH	Unit-V
	measurements	Bioinorganic
	and	chemistry:
	conductance	Metals in
	measurements.	biology-bulk
	Unit-IV	and trace
	Isomerism in	metals-
	Co-ordination	Structure and
	compounds:	function of
	Structural	Metalloporphyr
21,	isomerism:	ins-
The state of the s	Ionisation,	Chlorophyll-
	Hydrate and	Vitamin B12-
	Linkage	Structure and
	isomerism.	Biological role
74	Stereo	of Myoglobin
	isomerism:	and
	Geometrical	hemoglobin.
	isomerism in 4	Metallo
	and 6 - Co-	enzymes-
-	ordination	carboxy
	complexes.	peptidase and
	Optical	peroxidase-
	isomerism in 4	sodium and
	-and 6- Co-	potassium ion
	ordination	pump.Biologica
	compounds, 56	I functions and
	IV Applications	toxicity of
	of complexes in	chromium,
	quantitative	manganese,
1 2	analysis:	cobalt, nickel,
	Estimation of	copper, arsenic,
	Ni2+ using	iodine and

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1						1	
Ţ.					DMG and	mercury.	
	メサー	- Charles			Mg2+ using		
					Oxine.		
		e, interpretation	-47 . 2 . 2 . 1 . 1 . 1 . 1 . 1 . 1 . 1 . 1		Unit-V		
	77,30	195 A 4.1			Semiconductors		
	100	0 1 1 1	7.1		: Intrinsic and	÷ .	
		100 E.A.			Extrinsic Semi		
		haritar			Conductors - n-		
					type and ptype -		
	13 4 7 4				Properties and		
					uses.	50 d	
	, ii . 5	Magazini .	kala d	. 1.13	Structures of	Name of State (
		egir, e july,eren			alloys:		
		Cart. N. S.	1864. ** * *		Interstitial,		
		to the property of	g) = 4		substitutional	y 9 j	
	r Divisi	CARAGO INC.	160		and		
			11.30		intermetallic		
	1.5	100	R.J.		alloys.	41 1 - 4	
	\$.	-1 2 4x X				UNIT-III:	
	~~					Analysis of	
		ST 1				carbohydrates.	
	727	metal w.	•			UNIT-II:	
		H. Tallow				Reagents in	
	ė –	t lead to the second				organic	
	, , , , ,	Core Paper - VI:			2 1	synthesis:Ozon	Course
15	20UCY508	Organic	Unit 1 - 5	5%	-	e, Osmium	Teacher
		Chemistry- I				tetroxide,Lithiu	Cachel
						mAluminium	
					- nl	hydride,	
,						sodium	
						borohydride	10 - 10
		بيخر	6			and Birch	
						reduction. /	
						1	

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						UNIT -III	
		* * * * * * * * *			UNIT- II	Determination	
					Determination	of Degree of	
					of solubility of	hydrolysis-	
-1					sparingly	indirect method	
					soluble salts.	and Electrical	
		la de la companya de		-	UNIT- III	Conductance	
					choice	method.	
					ofIndicators	UNIT –IV	
					UNIT- IV	Electrochemica	
		Core Paper – VII			Equilibrium	1	Course
16	20UCY509	: Electro	Unit 1 - 5	5%	constant of a	instrumentation	Teacher
		Chemistry		3 4	electrochemical	and techniques:	reaction
		A. A.	37 T		reaction,	Potentiostatic	
		RET. MIT			Determination	and	- 1
			37.		of transport	galvanostatic	
		The Country of	1 AF		number.	Experiments.	
			244 - A	E	UNIT -V	Cyclic	
	52	14-)			Cathodic	voltammetry,	
					protection,	chronoampero	
		1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1		3	modifying the		
					environment.	chronopotentio	
						metry.	
					Unit V:	Unit V:	
	10.15				Types of textile		, 3, — , — ,
					fibres: Natural	of fibres -	
		. 4:			Semisynthetic		Course
17	20UCY510	Core Paper- VIII	: Unit 1 - :	5 10%	and Synthetic		
1,	2000111	Dye Chemistry					
					Wool, Silk	fibres).	
			1		acetate, Viscos		
					rayon,	textile fibres	
					Tayon,		

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1			-	-		Contract of the contract of th	
					polyamides,	Cotton, wool,	
					polyacrylonitril	silk, nylon,	
					e andpolyester.	polyester and	
					Selection of	polyacrylamide.	
		The spine is not	de par T		dyes for	Natural fibres:	
			· [,1]		different fibres,	Physical,	
	1				Fastness	chemical	
					properties of	properties and	
					dyes.	uses of	
						cellulose fibre	
	*					(cotton),	
						protein fibre	
		47.17				(silk and wool).	
						Synthetic	
						fibres:	
						Preparation,	
	7 1-1				1,	properties and	
						Uses of Nylon	
		- 48	have a fine		2	6, Nylon (6,6),	
					A Jirai	Polyester, and	
	1 × 1 1 1		4			poly acryl	
		and the second				amide. Semi	
					+	synthetic fibres:	
						Rayon -	
						manufacture of	
					Terror	viscose rayon	
						and	
					- 1.6. 9=	cuprammonium	× 4
					= 210	rayon.	
		Core Elective I:		-			
18	20UCY5E1		Unit 1 - 5	0%	-	-	-
		1					
19	20UCY5S1		Unit 1 - 5	0%	•	- /	-
18	20UCY5E1	Analytical Chemistry			-	-	7

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20	20UCY5S2	Information security Skill Based Elective –I : Cyber Security- Ethical Hacking	Unit 1 - 5	0%	-	Unit III: Peak area and proton counting.	-
21	20UCY611	Core Paper – IX: Physical Methods and Chemical Structure	Unit 1 - 5	5%	Unit III: Application of NMR in the study of simple molecules : xylene	Application of NMR in the study of simple molecules :diethyl ether Unit IV: Mass Spectra: Theory, basic principle, isotope abundance, base peak, molecular ion peak, Determination of molecular ion peak, meta stable ion, McLafferty rearrangement and Nitrogen	

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22 20UCY612 Core Paper – X : Core Paper – X : Organic Chemistry-II Chemistry-II Constitution and Synthesis of the following: Vitamin Al (retinol), Vit								
Terminal residual Analysis: N- Terminal end by Sanger's method. C- Terminal end by reduction Unit-IV Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y Constitution Course Teacher Course Teacher Course Teacher Course Teacher	(1)						rule.	
22 20UCY612 Core Paper – X: Organic Chemistry-II Unit 1 - 5 Chemistry-II Core Paper – X: Chemistry-II Core Paper – X: Chemistry-II Core Paper – X: Course Teacher Teacher Teacher Course Teacher	-						Unit-II	
Analysis: N- Terminal end by Sanger's method. C- Terminal end by reduction Unit-IV Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid				577			Terminal	
Core Paper – X: Organic Chemistry-II Unit 1 - 5 Chemical kinetics and photochemistry Core Paper – X: Core Paper – X: Organic Chemical kinetics and photochemistry Core Paper – X: Unit 1 - 5 O% Course Teacher Course Teacher Course Teacher Course Teacher Course Teacher				- 1- *y			residual	
22 20UCY612 Core Paper — X: Organic Chemistry-II Unit 1 - 5 Official kinetics and photochemistry y by Sanger's method. C- Terminal end by reduction Unit-IV Course Teacher			-	- PERFECT A			Analysis: N-	1.14
Core Paper – X: Organic Chemistry-II Unit 1 - 5 Office Paper – X: Organic Chemistry-II Unit 1 - 5 Office Paper – X: Organic Chemistry-II Unit 1 - 5 Office Paper – X: Organic Chemistry-II Chemistry-II Office Paper – X: Unit 1 - 5 Office Paper – X: Course Teacher			Mark Mark	70	- 4		Terminal end	
Core Paper – X: Organic Chemistry-II Unit 1 - 5 Off the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y Terminal end by reduction Course Teacher Course Teacher Course Teacher Course Teacher Course Teacher						1-17	by Sanger's	
22 20UCY612 Core Paper – X : Organic Chemistry-II Unit 1 - 5 0% - Unit-IV Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y mechanics							method. C-	
22 20UCY612 Organic Chemistry-II Unit 1 - 5 O% - Unit-IV Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y Course Teacher Teacher		***	1.0				Terminal end	* * *
22 20UCY612 Organic Chemistry-II Unit 1 - 5 0% Chemistry-II Unit 1 - 5 0% Chemistry-II Unit 1 - 5 0% Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y mechanics			Core Paper – X:				by reduction	C
Chemistry-II Constitution and Synthesis of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and photochemistr y mechanics	22	20UCY612	Organic	Unit 1 - 5	0%		Unit-IV	
of the following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and kinetics and photochemistr y mechanics			Chemistry-II	To be to			Constitution	leacher
following: Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and kinetics and photochemistr Quantum y mechanics			1.2				and Synthesis	
Vitamin A1(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and kinetics and photochemistr Quantum y mechanics				. * .	-		of the	
Al(retinol), Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and kinetics and photochemistr y mechanics			- 112				following:	
Vitamin B1, Pyridoxine and Ascorbic acid Chemical kinetics and kinetics and photochemistr y mechanics							Vitamin	
Pyridoxine and Ascorbic acid Chemical Chemical kinetics and photochemistr Quantum y mechanics				344	7		Al(retinol),	
Chemical Chemical kinetics and kinetics and photochemistr y mechanics			+		1		Vitamin B1,	3
Chemical Chemical kinetics and photochemistr Quantum y mechanics							Pyridoxine and	\$
kinetics and kinetics and photochemistr Quantum mechanics				4 -			Ascorbic acid	
photochemistr Quantum mechanics						Chemical	Chemical	
y mechanics			1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1				kinetics and	
	23	20UCY613				photochemistr	Quantum	
Core Paper - XI: Unit V Unit V			G 7 77	4-1			T	
					_	Unit-V	Unit-V	
Chemical Colloids: The need for Course								Course
23 200C Y 613 Kinetics and Unit 1 - 5 5% Classification, quantum				Unit 1 - 5	5%			
Quantum preparation and mechanics.								r, = 5
Mechanics application of Functions:			Mechanics					
colloids Real, complex,								9 4 4 1
Properties of odd, even,								
colloids: orthogonal and								
Optical and normalized						Optical and	normalized	

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1					kinetic	functions.	
					properties:	Operators:	
				-	Tyndall effect	linear and non-	
					and Brownian	linear,	75
	1 2				movement	differential,	
					Electrical	Hermitian,	
					properties:	Hamiltonian,	
					Charge on	momentum	
			_ 1		colloidal	(linear and	
					particle,	angular)	
	*		v '-		Electrical	commutator	
					double layer	(Theorems) and	
					and	non-	
-					Zetapotential.	commutators,	
					Electrokinetic	Eigen functions	
					properties:	and eigen	
					Electroosmosis	values.	
	7	0 · ×			and	Postulates of	
	- F	-27			Electrophoresis.	quantum	
	r Ligh				Emulsions:	mechanics	
					Preparation,	(statements	
					Properties and	only).	
	. 73				Applications.		
	1 7.3				Gels: Properties		
	,	5.000			and		
		in the second second			Applications.		
				<u>.</u>	UNIT-I	UNIT –II	
					Biodegradable	Miscellaneous	
-		Core Paper - XII:	77.72.1	504	Polymers:	Polymerization	Subject
24	20UCY614	Polymer	Unit 1 - 5	5%	Preparation,	reactions- Electrochemica	Expert
		Chemistry			Properties and application of		
					PolyLactic	Polymerization.	
		. 1 1 1	-		ToryEactic	, confinenzation.	

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1						T	11
					Acid.	UNIT-III	
					UNIT-V	Hydrolytic	
100					Injection	method	
					moulding, blow	UNIT –IV	3
			7.70 7.50		moulding	End group	
			W.			Analysis	
						UNIT -V	
	, a					Recent	
						Advances in	
				,		Polymers :	
						Polymer and	
						environmental	
						effect-	-
79.1						introduction-	
					**************************************	disposal of	
				1		polymer waste -	
						recycling	
	1.9.	e > 1				system-	w
						importance of	
						biopolymers.	
		Core Elective II-					
25	20UCY6E2		Unit 1 - 5	0%	-	1162	-
25	20001022	Chemistry – II					- V ₁ I
		Core Practical III:		<			1
		Gravimetric	Unit 1 - 5	0%			_
26	20UCY615	Analysis and	Omt 1 - 3	070			
		Physical					=
		Chemistry					
		Skill Based					
27	20UCY6S3	Elective –II :	Unit 1 - 5	0%	•	-	-
		Green chemistry	-				
-							/

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		Skill	Based				· jo
		Elective	–II :			4	
28	20UCY6S4	Theory	behind	Unit 1 - 5	0%	-	 -
		Practical			- F		
		chemistry				4	

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